# Chronic Kidney Disease of unknown etiology in Sri Lanka: An implication of Optimizing Recovery Ratio of Brackish Water Reverse Osmosis plant

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**Abstract**— Drinking water is an essential for human beings. The fresh water from the environment sources of water streams, wells and other water bodies are unable to use as drinking water because of the high content of dissolved salts and solids. This is most prominent areas where water is scarce and areas where fertilizers and chemicals are used for different day to day processes such as agriculture. With high level of environment contamination for a prolong period, the chemicals get into the underground water sources. In rural areas of Sri Lanka mainly in the district of Anuradhapura, majority of the humans are prone to Chronic Kidney Diseases of Unknown etiology (CKDu). Reverse osmosis technique is used to remove the dissolved solids from the fresh water obtained into a drinkable level. Brackish Water Reverse Osmosis (BWRO) plants are present in an industrial scale to provide drinking water from brackish water, but at a higher cost. The main aim of this project is to develop a BWRO plant at a lower operating cost with high recovery ratio to be implemented in areas where Total dissolved solid (TDS) levels are above the SLS standard (500 mg/L) and where even brackish water is scarce.

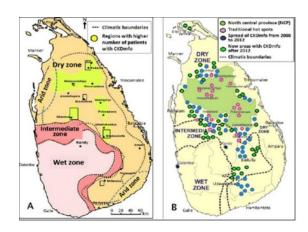
*Keywords*— Drinking Water, Brackish Water, Reverse Osmosis, Cost, Recovery Ratio

### I. INTRODUCTION

When considering the planet earth 70% of the area is covered with water bodies but 2.5% of it is fresh water which is suitable for human use and drinking purposes only 1% of that is easily accessible, other is saline water and ocean based. In most of the countries, lack of fresh water is a huge crisis. Even in the dry zone of Sri Lanka the freshwater crisis is inevitable with an average rainfall of 1000 mm (Wanasinghe et al., 2018) for a limited time period of 3 months and a drought period for 5 months during the mid-year. The water needed for the agricultural activities are obtained using the tank cascade system, which is to be sufficient for the drought period, once filled during the rainy season. But with time it has been observed that Dental fluorosis and Chronic Kidney Diseases of Unknown etiology (CKDu) is a major health issue as a noncommunicable disease mainly in parts of North Central, North, North Western, Central, Eastern, Uva, Southern (Dissanayake, 2020) which had result in high morbidity and mortality among farmers, agricultural laborer's and their families between the ages 40-60 due to the changes in the drinking water parameters by the contamination of

agricultural activities as a probable cause. The dialysis treatment and kidney transplantation are the only cure for the CKDu patients with a high cost or causing death within two years of getting prone for 80% of the patients.

Figure 1: Areas of CKDu spread (Wanasinghe et al., 2018)



The Brackish Water Reverse Osmosis (BWRO) is the process of obtaining purified drinkable fresh water by reverse osmosis through a specially designed semipermeable membrane comprising of pre-treatment, post treatment and a cleaning cycle. The feed water temperature, concentration, flow rate and pressure difference in sides of the semi-permeable membrane are the major factors effecting the process. Reverse osmosis is a technique that mainly utilizes to convert sea water into fresh water, on board naval vessels to meet the freshwater requirement. The Government of Sri Lanka initiated a program to supply safe drinking water CKDu affected people by importing BWRO plants through National Water Supply and Drainage Board (NWSDB) of Sri Lank, Non-Governmental Organizations (NGO) and establishment of distribution stations in the North Central Province (NCP). Amidst of the failure of that program due to practical reasons of unavailability of skilled personnel for defect rectification, repairs and absence of compatible spare parts in the local market. Then, with the involvement of the Sri Lanka Navy, a sustainable solution of in-house BWRO plants was introduced by utilizing SLN operators to supply safe drinking water to the rural community (Dissanayake, 2020). Consequently, SLN BWRO plants project has being expanded up to 900 plants and providing safe drinking water to more than 2 million people at CKDu impacted areas in the country.

The absence of drinking water quality parameters within acceptable levels (SLS 614: 2013) is the main cause for the

spread of CKDu in Sri Lanka. Elimination of CKDu can be done by establishment of BWRO plants. But the Recovery ratio of existing BWRO plants is calculated and identified as around (40-50) % of feed water (Indika et al., 2021). Further, it is understanding that present recovery is not adequate comparative to the ground water yield at the dry zone in Sri Lanka. A 75% recovery ratio is a mandatory requirement of a BWRO plant as for the United States Environmental Protection Agency (EPA) due to the scarcity of fresh resources.

#### II. DESIGN FEATURES

North Central Province (NCP) is the most CKDu affected area in the country. Therefore, 407/ Meegassegama village, Talawa divisional secretariat, Anuradhapura District of the NCP was selected as the study area in this research. The process would differ according to the implementation area because the quality of brackish water, contamination levels and total dissolved solid (TDS) levels differ from area to area. Then the feed water characterization should be done for the identification of the TDS levels in the feed water for the membrane selection process. In the BWRO process to increase the efficiency by determining the membrane design calculating the effective operative feed water inlet pressures at each stage and the flow rate. The input pressures of pumps should be calculated with trial and error through the closed loop monitoring of the output and inputs by considering the head losses. The whole process is to be automated as PID to control input, output pressures, with the operation of directional valves to maintain constant flowrate, freshwater feed system for membrane flushing at the starting process. Implementation area was selected as for the requirement of a low-cost brackish water reverse osmosis (BWRO) plant with the feed water characteristic carried out. The observation of the existing BWRO system with two pressure vessels arranged as single stage parallel configuration was done by measuring the output flows of permeate and concentrate, manually using measuring instruments. The main challenge of coming up with the membrane arrangement was discussed. Mainly the stage array module arrangement and the closed-circuit desalination (CCD) were the most suitable way used to increase the recovery ratio as well as to increase the permeate water quality.

The closed-circuit desalination (CCD) involved high number valves and pumps, hence our main objective is to improve the recovery ratio at a lower cost, by the trials carried out on the test rig the concentrate staging array system with two modules had a high recovery ratio. The readings for the output flows of permeate and concentrate was taken manually but even with high pressure pump turned on the recovery ratios were very low due to the absence of the required osmotic pressure for the RO membrane. The osmotic pressure is developed by using

flow restrictors.

Feed Water Characterization

A feed water characterization of water samples taken from the implementation area was done to identify the water quality in the area.

Following details were obtained through the water quality tests which were conducted.

National Water Supply and Drainage Board - Water quality test report

Date and time of water sample collected : 0900 hrs on 2020.08.04

Date and time of water sample handed over : 1025 hrs on 2020.08.04

Location of water sample : 407, Meegassegama, Kiralogama

Table: 1 Feed Water Characteristic

Parameter s         614: 2013 standards Maximu m Value         Relevan t to the Sample         Method of Testing           Physical Characteristics           Colour         Hazen         15         15         APHA 2120 C           Turbidity         NTU         2         3.0         APHA 2130 B           Physical Characteristics           Turbidity         NTU         2         3.0         APHA 2130 B           APHA 2130 B           Chemical Characteristics           TDS         mg/l         500         227         APHA 2540 c           Chloride (Cl-)         mg/l         250         70         APHA 4500 classes (as cacoa)           Total bases (as CaCO3)         mg/l         200         140         APHA 2320 B           Nitrate (as NO3)         mg/l         250         130         APHA 4500 classes (APHA 4500 classes)           Nitrite (as NO3)         mg/l         3         0.01         APHA 4500 classes (APHA 4500 classes)           Nulphate (as $SO_4^{2-}$ )         mg/l         250         04         APHA 4500 classes)           Sulphate (as $SO_4^{2-}$ )         mg/l         250         04         APHA A2500 classes)           Fluoride         mg/l         1					1 1
S         Units         Maximu m Value         t to the Sample         d of Testing           Physical Characteristics           Colour         Hazen         15         15         APHA 2120 C           Turbidity         NTU         2         3.0         APHA 2130 B           pH value $25^{\circ}$ C $\pm$ 0.005°C         6.5-8.5         7.3         APHA 4500 H* B           Chemical Characteristics           TDS         mg/l         500         227         APHA 2540 c           Chloride (Cl-)         mg/l         250         70         APHA 4500 Cl -B           Total bases (as CaCO <sub>3</sub> )         mg/l         200         140         APHA 2320 B           Total Solubility (as CaCO <sub>3</sub> )         mg/l         250         130         APHA 2340 C           Nitrite (as $NO_3^-$ )         mg/l         50         0.44         APHA 4500 $NO_3^-$ , E           Nitrite (as $NO_2^-$ )         mg/l         3         0.01         APHA A500 $NO_2^-$ B           Sulphate (as $SO_4^2^-$ )         mg/l         250         04         APHA A500 $SO_4^2^-$ B           Fluoride         mg/l         1.0         0.42         APHA A500 APHA APHA APHA APHA APHA APHA APHA AP	Demonstra	According to SLS			Metho
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Nitrate (as $NO_3^-$ )         mg/l         50         0.44         APHA 4500 $NO_3^-$ , E           Nitrite (as $NO_2^-$ )         mg/l         3         0.01         APHA 4500 $NO_2^-$ B           Sulphate (as $SO_4^{2-}$ )         mg/l         250         04         APHA 4500 $SO_4^{2-}$ B           Fluoride         mg/l         1.0         0.42         APHA 4500 $SO_4^{2-}$ B	`	111g/1	230	130	2340 C
Nitrate (as $NO_3^-$ )         mg/l         50         0.44         4500 $NO_3^-$ , E           Nitrite (as $NO_2^-$ )         mg/l         3         0.01         APHA $4500$ $NO_2^-$ B           Sulphate (as $SO_4^{2-}$ )         mg/l         250         04         APHA $4500$ $SO_4^{2-}$ B           Fluoride         mg/l         1.0         0.42         APHA $4500$ $SO_4^{2-}$ B	CaCO <sub>3</sub> )				
$NO_3^-$ )         mg/l         50         0.44         4500 $NO_3^-$ )         Moganization         APHA           Nitrite (as $NO_2^-$ )         mg/l         3         0.01         4500 $NO_2^-$ B         APHA         4500         APHA           (as $SO_4^{2-}$ )         mg/l         250         04         4500 $SO_4^{2-}$ B         APHA           Fluoride         mg/l         1.0         0.42         4500	Nitrata (ac				APHA
Nitrite (as $NO_2^-$ )       mg/l       3       0.01       APHA $4500$ $NO_2^-$ B         Sulphate (as $SO_4^{2-}$ )       mg/l       250       04       APHA $4500$ $SO_4^{2-}$ B         Fluoride       mg/l       1.0       0.42       4500 $SO_4^{2-}$ B	,	mg/l	50	0.44	
Nitrite (as $NO_2^-$ )       mg/l       3       0.01 $4500$ $NO_2^-$ B       Sulphate (as $SO_4^{2-}$ )       mg/l       250       04       APHA $4500$ $SO_4^{2-}$ B       APHA         Fluoride       mg/l       1.0       0.42       4500         APHA	$NO_3$ )				$NO_3^-$ , E
$NO_2^-$ )     mg/l     3     0.01     4500 $NO_2^-$ B       Sulphate (as $SO_4^{2-}$ )     mg/l     250     04     4500 $SO_4^{2-}$ B       Fluoride     mg/l     1.0     0.42     4500       APHA       4500	Nitrita (ac				APHA
Sulphate (as $SO_4^{2-}$ ) mg/l 250 04 APHA 4500 $SO_4^{2-}B$ Fluoride mg/l 1.0 0.42 4500		mg/l	3	0.01	4500
Sulphate (as $SO_4^{2-}$ )         mg/l         250         04         4500 $SO_4^{2-}B$ Fluoride         mg/l         1.0         0.42         4500 $SO_4^{2-}B$	NO <sub>2</sub> )				$NO_2^-B$
(as $SO_4^{2-}$ ) mg/l 250 04 4500 $SO_4^{2-}B$ Fluoride mg/l 1.0 0.42 4500	Sulphoto				APHA
Fluoride $mg/l$ 1.0 0.42 4500		mg/l	250	04	
Fluoride mg/l 1.0 0.42 4500	$(as 30_4)$	_			$SO_4^{2-} B$
$\frac{1}{1}$	Elucai de				
(og F-)   IIIg/I   1.0   0.72   4300	(as F <sup>-</sup> )	mg/l	1.0	0.42	4500
$(as F)$ $F^-D$	(as r)				$F^-D$
Total APHA					APHA
Phosphate mg/l 2.0 0.28 3500	Phosphate	mg/l	2.0	0.28	3500
$(as PO_4^{3-})$ PE					PE

Iron (as Fe)	mg/l	0.3	0.09	APHA 3500- Fe B
Free Ammonia (as NH <sub>2</sub> )	mg/l	0.06	-	PAN Metho

Sample Drawn by S and D Chemicals (PVT) Ltd.

Test on Heavy Metals

Sample Received date : 21/09/2020

Report Issued Date : 21/09/2020

Table 2: Heavy Metal Indication

Parameter	CAS Number	Result(mg/l)	Limit of Detection (mg/l)
Arsenic as As	7440-38- 2	ND	0.05
Cadmium as Cd	7440-43- 9	ND	0.05
Copper as Cu	7440-50- 8	ND	0.05
Chromium as Cr	7440-47- 3	ND	0.05
Iron as Fe	7439-89- 6	ND	0.05
Lead as Pb	7439-92- 1	ND	0.05
Magnesium as Mg	7439-95- 4	8.51	0.10
Manganese as MN	7439-96- 5	ND	0.05
Nickel as Ni	7440-02- 0	ND	0.05
Sodium as Na	7440-23- 5	27.93	0.10
Potassium as K	7440-09- 7	0.26	0.10
Calcium as Ca	7440-70- 2	31.78	0.10
Silver as Ag	7440-22- 4	ND	0.05

Pressure Vessel Arrangement

Initially the pressure vessel/membrane arrangement, which was in two stage parallel configuration, where the water from the pre-treatment stage is inlet into two parallel pressure vessels, the concentrate and the permeate output is also obtained parallelly from the two pressure vessels was modified and tested on the BWRO test rig present at the marine laboratory of KDU as a two stage series arrangement, where the water from the pre-treatment is inlet into one pressure vessel, the output concentrate from the first vessel is inlet into the second pressure vessel and both permeates from first stage and second stage are combined as the final output. Following reading were

obtained without controlling the pressure of the concentrate output using a globe valve.

Figure 2:Pressure vessel arrangement before modification

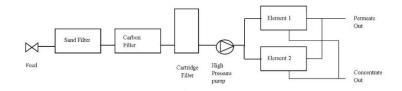


Figure 3: Pressure vessel arrangement after modification

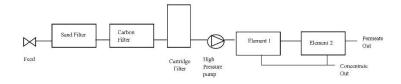


Table 3: Results obtained without concentrate pressure control

Descri	ption	Perme ate Out (ml)	Conc entra te Out (ml)	Feed water (ml)	Recovery Ratio (%)
Befor e Modif	Without pressure pump	125	4000	4125	3.030303
icatio	With pressure pump	140	4000	4140	3.381643
After Modif	Without pressure pump	350	4000	4350	8.045977
icatio n	With pressure pump	600	4000	4600	13.04348

The concentrate output needs to be controlled using a globe valve to build up a pressure at the output of the pressure vessel in order to limit the concentrate flowrate and increase the permeate flowrate increasing the overall recovery ratio per fixed amount of inlet feed water.

## **Recovery Ratio Calculation**

The calculation of the recovery ratios for the above module Arrangements are as follows.

- $Qfn = \text{Feed flow to RO-n } (m^3/h)$
- Qpn = Permeate flow of RO-n
- Qbn = Concentrate flow of RO-n
- R = Recovery ratio

Single stage module arrangement

$$Qf1 = Qp1 + Qbl$$

$$R = 100 * \frac{Qp1}{Qf}$$

Multistage parallel module arrangement

$$Qf1 = Qp1 + Qp2 + Qp3 + Qbl + Qb2 + Qb3$$

$$R = 100 * \frac{Qp1 + Qp2 + Qp3}{Qf1}$$

Multistage series module arrangement

$$Qf1 = Qp1 + Qbl$$

$$Qf2 = Qp1 = Qp2 + Qb2$$

$$R = 100 * \frac{Qp2}{Qf1}$$

Multistage series and parallel module arrangement

$$Qf1 = Qp1 + Qbl$$

$$Qf2 = Qp1 = Qp2 + Qb2$$

$$Qf3 = Qb1 + Qb2 = Qp3 + Qb3$$

$$R = 100 * \frac{Qp2 + Qp3}{Qf1}$$

Permeate flow (Qp)

$$Qp = Am * Lp * (\Delta Pm - \sigma \Delta \Pi)$$

- Am = Total membrane surface area
- *Lp* = Hydraulic permeability of the membrane (depends on the type and product of the membrane)
- $\triangle Pm$  = Average transmembrane pressure
- $\sigma$  = Reflection coefficient of membrane.
- $\Delta \eta$  = Average osmotic pressure difference between feed  $\Delta \eta f$  and permeate  $\Delta \eta p$ .
- $(\Delta Pm \sigma \Delta \eta)$  = Average trans membrane net driving pressure, NDP.

Number of membrane elements (Ne)

Ne = 
$$\frac{Qp}{f * Se}$$

- Qp = Design permeate flow
- F = Flux
- Se = Membrane element surface area

Number pressure vessels (Nv)

$$Nv = \frac{Ne}{Nepv}$$

- Ne = Number of membrane elements
- Nepv = Number of elements in a pressure vessel
- The number elements and the number pressure vessels may vary with the manufacturer due to the variation of the Membrane element surface area (Se) and Number of elements in a pressure vessel (Nepv).

Implementation at Talawa (Anuradapura District)

The BWRO facility was built at 407/ Meegassegama village, Talawa divisional secretariat, Anuradhapura District of the North Central Province. There are 520 families living in this village.

Figure 4: BWRO Facility building



Membrane Installation

The existing BWRO was a single stage single element arrangement figure 6. Therefore, 2<sup>nd</sup> membrane was purchased and installed serially to this BWRO plant figure 8.

Salient Features of Membarne

: LP 440 Modle Working Pressure (Max) : 600 psi h Working Temperature of feed water (Max): 45 °C c. Operating PH d. : 2-11 Chlorine concentration : < 0.1 ppme. f. Maximum SDI : 5

The membrane installation is a very delicate procedure if processed wrongly would result in membrane damage and water leaks. First the flow side of the membrane should be identified by the marking then the pressure vessel end caps

must be adjusted to decide the orientation of the pressure vessel, to define the inlet port, permeate out and the concentrate out. When the inserting the membrane inside the pressure vessel, the end caps are fixed accordingly as for the end for the flow side which contains the 'O' ring to be placed at last to the vessel. Shampoo was applied as lubrication before inserting the element. Then the ½ inch sockets were placed at the feed inlet, permeate out, concentrate out ports and end cap to the remaining inlet port as there are 2 ports per end cap of the pressure vessel.

Figure 5: Inserting the new BWRO membrane to the pressure vessel



The tightening of the sockets at the end caps with appropriate amount of thread sealant and PVC gum is necessary to avoid leaks as the membrane is pressured at high pressures. The leaks would also create a drop in pressure and a messy environment around the plant, even creating possibilities for electrical short circuiting. The locking mechanism of the end caps were different from what we have come across. The existing connections were observed and the flow rate of the concentrate was measured before removing for modification figure 6 & 7.

Figure 6: Design before modification

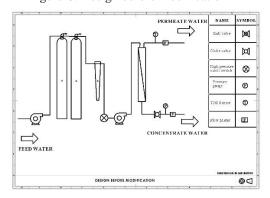


Figure 7: Observation of single stage arrangement



Figure 8: Time taken to fill 4L of concentrate with single stage membrane arrangement



As seen above, only 19.5s was taken to fill 4L amount of concentrate, our main objective was to increase. Which would decrease the flow rate and the amount of concentrate output, simultaneously increasing the permeate output. The union joints at the feed inlet, permeate output and the concentrate output were detached and the connections were modified to support the new additional membrane to the system to be modified as two stage series configuration. In addition to the plumbing the frame structure was modified to hold the second pressure vessel. There was a difficulty of attaching second pressure vessel as there was no compatible brackets, but the team was able to adjust an existing bracket to able to fit the new membrane while supporting the lower part via a temporary fixture, till a suitable bracket is available for installation in figure 9 & 10.

Figure 9: Modified design

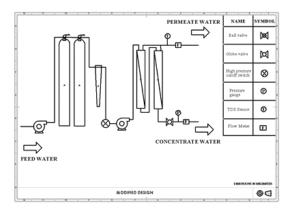


Figure 10: Modifying the plumbing arrangement



Same time, rejected water was tested by hand held conductivity meter and pH meter in Table 4.

Table 4: Comparison of Reject Water Parameters vs Sri Lankan Standard 614

Parameters	Results	SLS 614
pН	7.3	6.5-8.5
Electrial Conductivity	918 ppm	750 ppm
TDS	588 mg/l	500 mg/l
Hardness	518 mg/l	250 mg/l

### III RESULTS AND DISCUSSION

## Recovery Ratio

The membranes can be used relatively a longer duration than usual by performing cleaning cycles in a pre-defined period, chemically and fresh water back washing. If the process is slowed down or if the recovery ratio is decreased due to membrane clogging, the manufacture has provided with option of using the membrane in reverse flow direction by using the "O" Ring in the opposite side groove. The modification of the pressure vessel arrangement was successful in increasing the recovery ratio more than 50% which was present before with the single stage array arrangement by two stage series array arrangement. But the desired 75% recovery ratio was not obtained. The input and the output of the pressure pump was by-passed with a ball valve by creating a pressure difference and a flow path to supply for the water deficit created at the stage array input figure 11 & 12. Then the flow rate of the concentrate was as to fill 4L time occurred was 43.5s. Where the flowrate has decreased comparative to the 19.5s occurred previously with the use of single stage system. The resultant concentrate water also had more salt texture relative to the concentrate obtained before the modification figure 8.

Figure 11: Modified design with by-pass valve

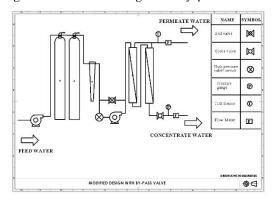


Figure 12: By-Pass valve modification

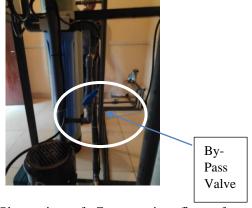


Figure 13: Observation of Concentration flow after modification





When calculating the recover ratio, we considered a time of 1 minute for calculations.

For a 1-minute time.

Feed water volume = 25.261

Before Modification,

Concentrate Volume = 12.91

Therefore, Permeate volume = 25.26 - 12.9 = 12.361

Recovery Ratio =  $12.36/25.26 \times 100\% = 48.9\%$ 

After Modification,

Concentrate Volume = 6.21 1

Therefore, Permeate volume = 25.26 - 6.21 = 19.05 1Recovery Ratio =  $19.05/25.26 \times 100\% = 75.4\%$ .

pH value of reject water was 7.3 and revealed that pH value was withn SLS 614. Further, TDS and Hardness of reject water was above specified SLS 614. Moreover, electrical conductivity was bit high compare to SLS 614.

### IV. CONCLUSION

This resulted in a huge benefit to the villages of Meegassegama by increasing recovery ratio which would in return reduce the cost per litre of water obtained and the reduce in feed water wastage through concentrate. The feed water saving would be mostly affected on times of drought where water is scarce, and the feed water sources are drying

Recommend to reuse reject water and improve the recovery ratio upto 100% Table 4.

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#### AUTHOR BIOGRAPHIES



Cmde (E) MCP Dissanayake, CEng (India) is currently performs as the Head of Department (Marine Engineering) and hold 2 No's patents for his research papers published so far. He is an inventor and published 06 No's publications on Brackish Water Reverse Osmosis application, Fan Boat Building and Oscillation Water

Column, Ocean Wave Eenrg Converter. He was the Director in Research & Development at Sri Lanka Navy and has received commendations in number of occasions from Commander of the Navy, HE the President of Sri Lanka for his innovation. Further, he was awarded with prestigious, Japanese, Sri Lanka Technical Award for his own developed low-cost Reverse Osmosis Plant, to eliminate Chronic Kidney Disease from Sri Lanka. Moreover, he has a vast exposure on marine diesel engines and possesses a Masters degree in Marine Engineering from Autralian Maritime College, University of Tasmania, Australia.